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An acid dissociation constant,  $K_a$ , (also known as acidity constant, or acid-ionization constant) is a quantitative measure of the strength of an acid in solution. It is the equilibrium constant for a chemical reaction known as dissociation in the context of acid-base reactions. In aqueous solution, the equilibrium of acid dissociation can be written symbolically as:

### **Acid dissociation constant - Wikipedia**

Nicholas Georgescu-Roegen (born Nicolae Georgescu, 4 February 1906 – 30 October 1994) was a Romanian American mathematician, statistician and economist. He is best known today for his 1971 magnum opus on The Entropy Law and the Economic Process, in which he argued that all natural resources are irreversibly degraded when put to use in economic activity.

### **Nicholas Georgescu-Roegen - Wikipedia**

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### **Fundamentals of Electric Circuits: Charles K Alexander**

I tried MealSquares a while back, based on the sidebar advertisement, and liked them pretty well

(â€œlikedâ€• in the â€œthis works well for meâ€• sense, not the â€œfood rewardâ€• sense).

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### **Hydraulic Power Pack: This Ebook Answer You All Questions**

Lull, William P., with the assistance of Paul N. Banks. Conservation Environment Guidelines for Libraries and Archives. Ottawa, ON: Canadian Council of Archives, 1995. National Information Standards Organization. Environmental Guidelines for the Storage of Paper Records. Technical Report NISO-TR01-1995.

### **References â€“ Heritage Conservation Info**

2 UNCG Undergraduate Bulletin 4 Notices Equality of Educational Opportunity The University of North Carolina at Greensboro is com-mitted to equality of educational opportunity and does not

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