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Table 9: 4-1 Electrolytes. Table 10: 5-1 Electrolytes. Table 11: 2-2 Electrolytes. Appendix 4.5 "Pitzer Parameter Derivatives. Table 1: Temperature Derivatives of Parameters for 1-1 Electrolytes Evaluated from Calorimetric Data. Table 2: Temperature Derivatives of Parameters for 2-1 and 1-2 Electrolytes Evaluated from Calorimetric Data.

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For these electrolytes the concentration c corresponding to the mass percent values given here can be found in the table "Concentrative Properties of Aqueous Solutions" in Section 8. references 1. CRC Handbook of Chemistry, and Physics, 70th Edition, Weast, R. C., Ed., CRC Press, Boca Raton, FL, 1989, p. D-221. 2. Wolf, A.

electrical Conductivity of aqueous solutions references

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An electrolyte is a compound which produces an ionic solution when dissolved in an aqueous solution. For example, a salt like KCl would produce an electrolyte solution. Those compounds which produce a large number of ions in solution are ... produce electrolyte solutions too. Common examples are acids produced by

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for some common electrolytes in aqueous solution at concentrations up to 0.1 mol/L. The units of \hat{V} are $10^{-4} \text{ m}^3 \text{ mol}^{-1}$. For very dilute solutions, the equivalent conductivity for any elec- ... CRC Handbook of Chemistry and Physics, 91th Edition Author: Dr. David Lide

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